



## **31. PV-Symposium Staffelstein 09.-11. März 2016**

**Performance und Sicherheit von  
PV-Li-Speicherbatterien**

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Baden-Württemberg (ZSW)**

# Battery Research at ZSW

## Materials Research – Cell Design – Post Mortem Analysis

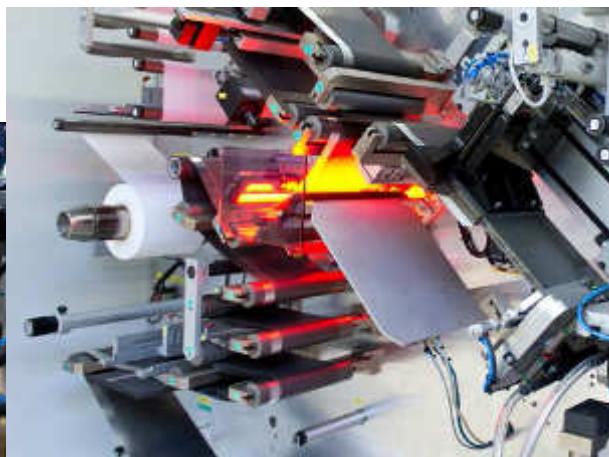
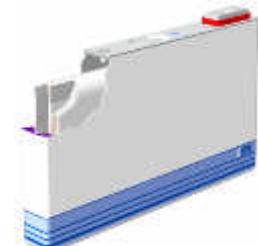
- Synthesis and characterization of new electrode materials
- Materials development including beyond lithium
- Process development for electrode and cell preparation
- Validation of new cell chemistries in reference cells
- Ageing mechanisms and post mortem analysis



# Battery Research at ZSW

## FPL - Research Platform for Production of large Lithium Ion Cells

- Fully automatic slurry preparation 60 l batches
- Double side electrode coating (500 mm width, 20 m/min)
- Fully automatic winding, assembling, electrolyte filling
- Formation 240 cycling and  
1.920 storage places



# Battery Research at ZSW

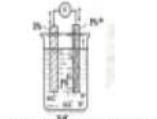
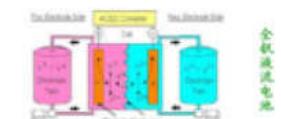
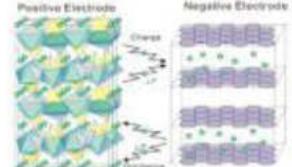
## Performance Test – Safety Test – BMS

Test field for cells, modules and battery systems

- Lifetime and performance testing
- Safety (abuse) test centre
- Battery management system and monitoring technologies
- Mathematical modelling and system engineering



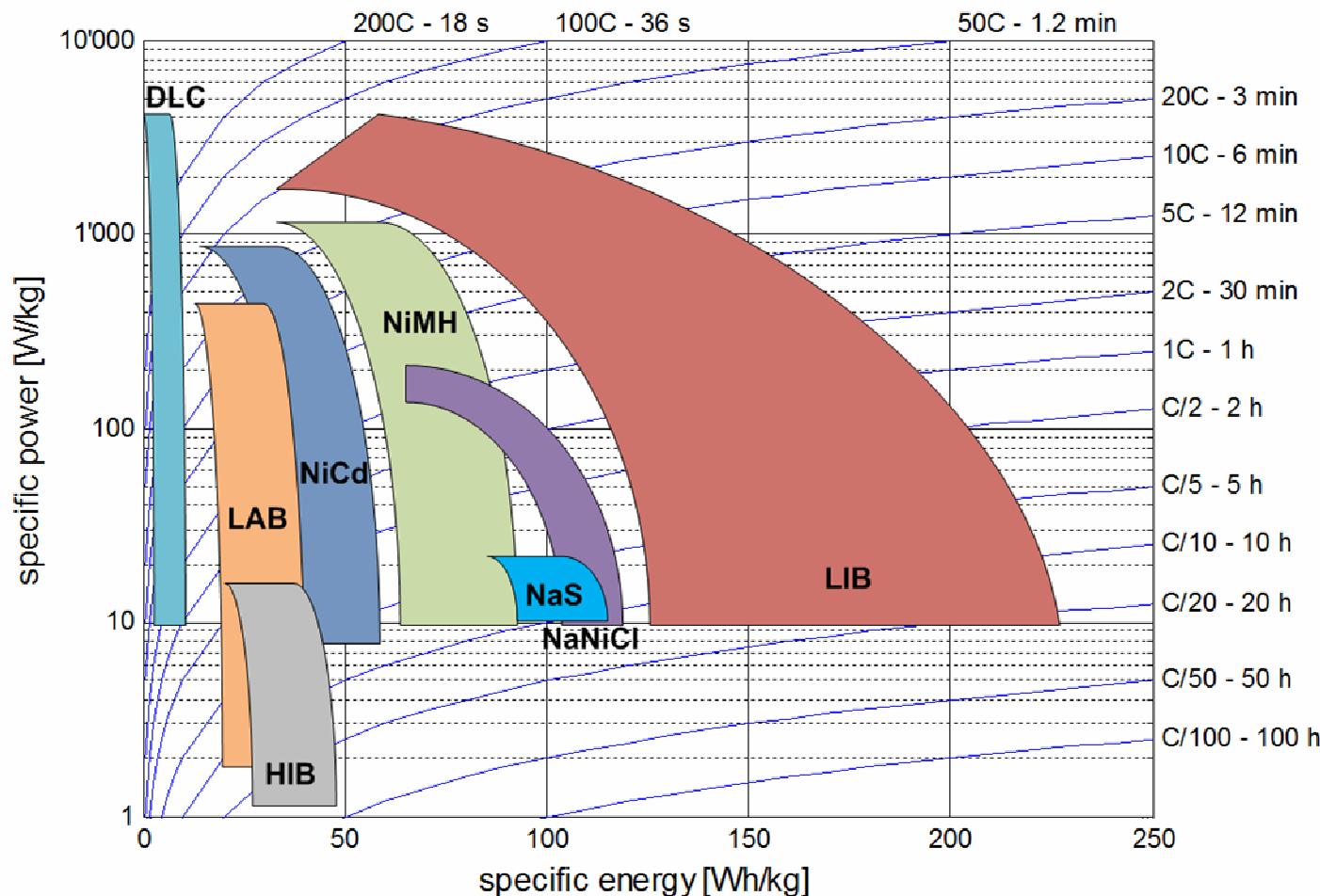
# Performance different storage systems

	Pumped Hydro	Lead-Acid	VRB	Na-S	BYD LFP
<b>Principle</b>	 $\text{Zn}^{2+} + \text{FePO}_4^{\text{2-}} + \text{TiO}_2^{\text{2-}} \rightleftharpoons \text{ZnFePO}_4^{\text{2-}} + \text{Ti}^{2+} + \text{H}_2\text{O}$ 正极: $\text{PbO}_2 + 2\text{H}^+ + 2e^- \xrightarrow{\text{放电}} \text{Pb} + \text{H}_2\text{O}$ 负极: $\text{V}^{2+} + e^- \xrightarrow{\text{放电}} \text{V}^{3+}$	 $\text{PbO}_2 + 2\text{H}^+ + 2e^- \xrightarrow{\text{放电}} \text{Pb} + \text{H}_2\text{O}$ $\text{Pb} + \text{H}_2\text{O} \xrightarrow{\text{充电}} \text{PbO}_2 + 2\text{H}^+ + 2e^-$	 全钒液流电池 正极: $\text{VO}_2^{+} + 2\text{H}^+ \xrightarrow{\text{放电}} \text{VO}^{2+} + \text{H}_2\text{O}$ 负极: $\text{V}^{2+} + e^- \xrightarrow{\text{放电}} \text{V}^{3+}$	 钠硫电池 正极: $2\text{Na}^{+} + 2e^- \xrightarrow{\text{放电}} 2\text{Na}^{+} + 2e^-$ 负极: $\text{S}^{2-} + 2e^- \xrightarrow{\text{放电}} \text{S}^{2-}$	 Positive Electrode: $\text{LiFePO}_4$ Negative Electrode: Graphite
<b>Response time</b>	10s (10% load)	100ms level	100ms level	100ms level (during running)	100ms level
<b>Energy Density (Wh/kg)</b>	/	35	10	170	100~130
<b>Round-trip Eff.</b>	70~80%	65~70% (battery 75%)	70~75% (battery 80%)	85% (during running)	85~92% (battery 97%)
<b>Temperature (°C)</b>	No icing	-15~40	10~50	310~350	-20~60
<b>Cycle Life</b>	50 years	1,000(75%DOD)	10,000(75%DOD)	3,000(75%DOD)	10,000(75%DOD) 6,000 (100%DOD)
<b>Investment (\$/kWh)</b>	200*	250*	/	700*	500 (Battery 400)

Quelle: Vortrag BYD

# Performance different batteries RAGONE

DLC	Doppelschichtkondensator	NiMH	Nickel-Metallhydrid-Batterie
LAB	Blei-Säure-Batterie	NaS	Natrium-Schwefel-Batterie
HIB	Hybrid-Ionen-Batterie (Aquion)	NaNiCl	Natrium-Nickel-Chlorid-Batterie (ZEBRA)
NiCd	Nickel-Cadmium-Batterie	LIB	Lithium-Ionen-Batterie



# Performance different batteries

Technology	Advantage	Disadvantage
pumped hydro power station	established technology low cost, long life	Long construction time, large space, large invest, requires special topography
<b>Lead acid</b>	Low cost, easy recycling good discharge ability, robust, wide spread production facilities, safety	Low energy density, short cycle life, poor low SOC performance, gas emission requires special rooms
Redox flow (V)	simple design High safety	Complex system, low energy/power density, leakage problems, high cost,
NaS, Na/NiCl <sub>2</sub>	Good energy density,	Permanent high temperature (losses), Complex BMS
<b>Li - Ion</b>	High power and energy density	Safety, complex BMS

## Performance Lead Acid vs. Li-Ion

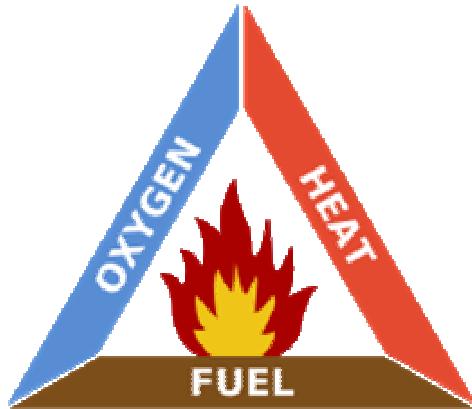
	Lead acid	Li-ion
Power ability	+	++
specific energy Wh/kg	0	++
energy density Wh/lit	+	++
efficiency/self discharge	+	++
Cycle life	0	+ (+)
cost	++	0
BMS	++	-
Operation temperature	+	+
Installation environment	0 (+)	++
Safety/failure tolerance	++	- -
recycling	++	-

# The Safety Hazard of Li-Ion Batteries

## Oxidant

### Delithiated Cathode:

- High oxidation potential
- Heat generation and gas formation
- linked with O<sub>2</sub> release



### Internal triggers (short circuit):

- cell balance, converting (Li-plating)
- depth discharge (Cu-plating)
- manufacturing defected separator
- Separator: melting
- deformation/ shock/ vibration

## Trigger



## Gas Formation

Electrolyte- Oxidation  
and Reduction

## Fuel

### Electrolyte:

- low flashpoints of the solvent (DMC= 25°C, EC= 160°C)
- exothermic decomposition

### Lithiated Anode:

- high reduction potential,
- decomposition reactions (>100°C)
- heat and gas formation (CO, H<sub>2</sub>)

### External triggers:

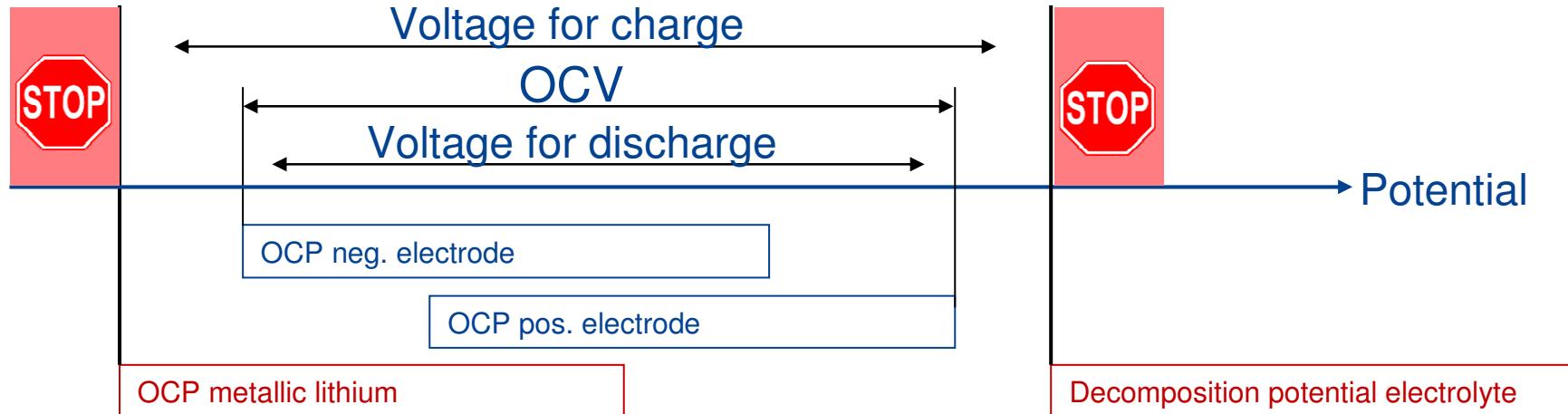
- overcharge
- external heat
- external short circuit

## Heat Formation

Exothermic  
Decomposition



# Operation limits for Li-Ion- batteries



- Limitation of the operation voltage window and limitation of the current (in particular charge current)
- No tolerance against overcharge and over discharge (deep discharge)

## Temperature limits

- Comfortable range               $10 - 30 \text{ }^{\circ}\text{C}$
- Usual operation window         $0 - 50 \text{ }^{\circ}\text{C}$
- Higher temperatures: accelerated ageing, safety
- Lower temperatures risk for Li plating while charging

# Overcharge

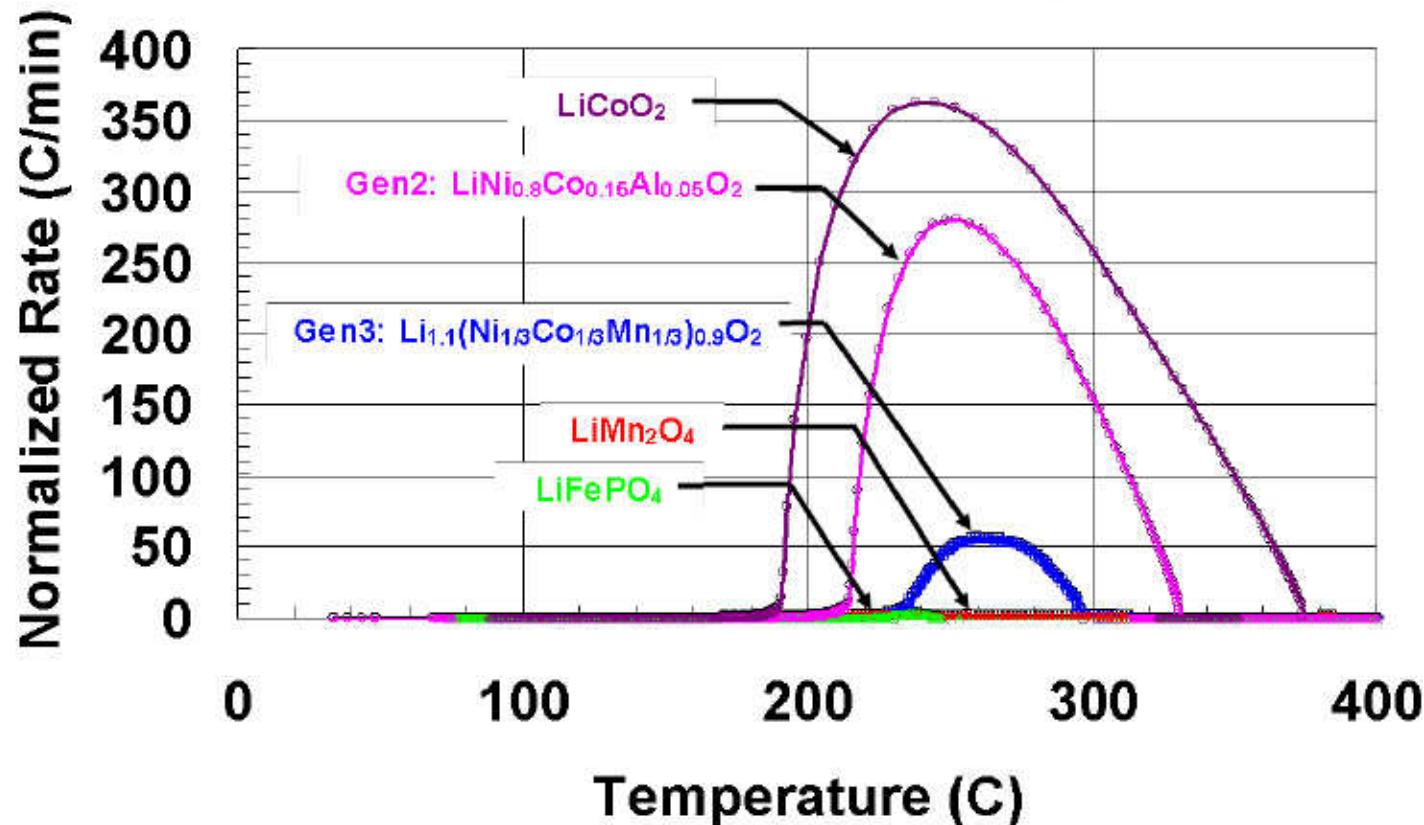


Delay in ignition of the emitted gases might lead to explosion

# Safety issues of materials

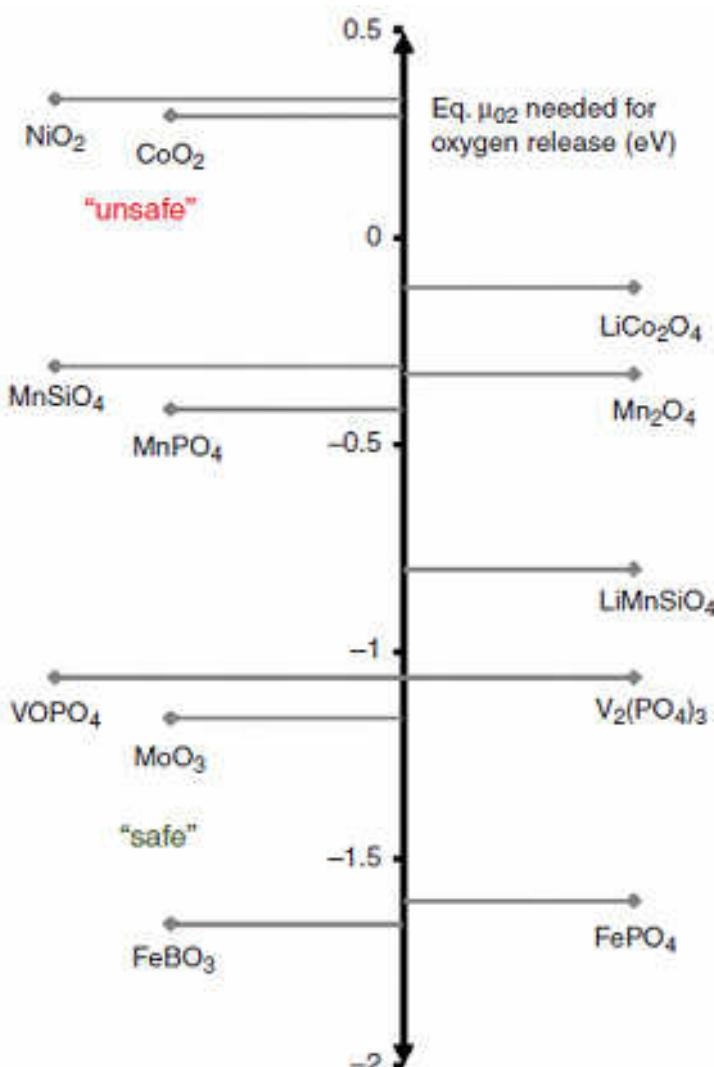
## Cathode

- source of heat formation
- different onset and amount of heat generation
- oxygen formation



Roth et al., Pacific Power Sources Symposium 2009

# Classification of Cathode Materials by their Oxygen Chemical Potential



thermal unstable



thermal unstable at  $> 500^\circ C$



the probability of oxygen formation decreases the safety

increasing thermal stability:



[Cider, MRS Bulletin 35 (2010) 693]

# Safety issues of electrolyte

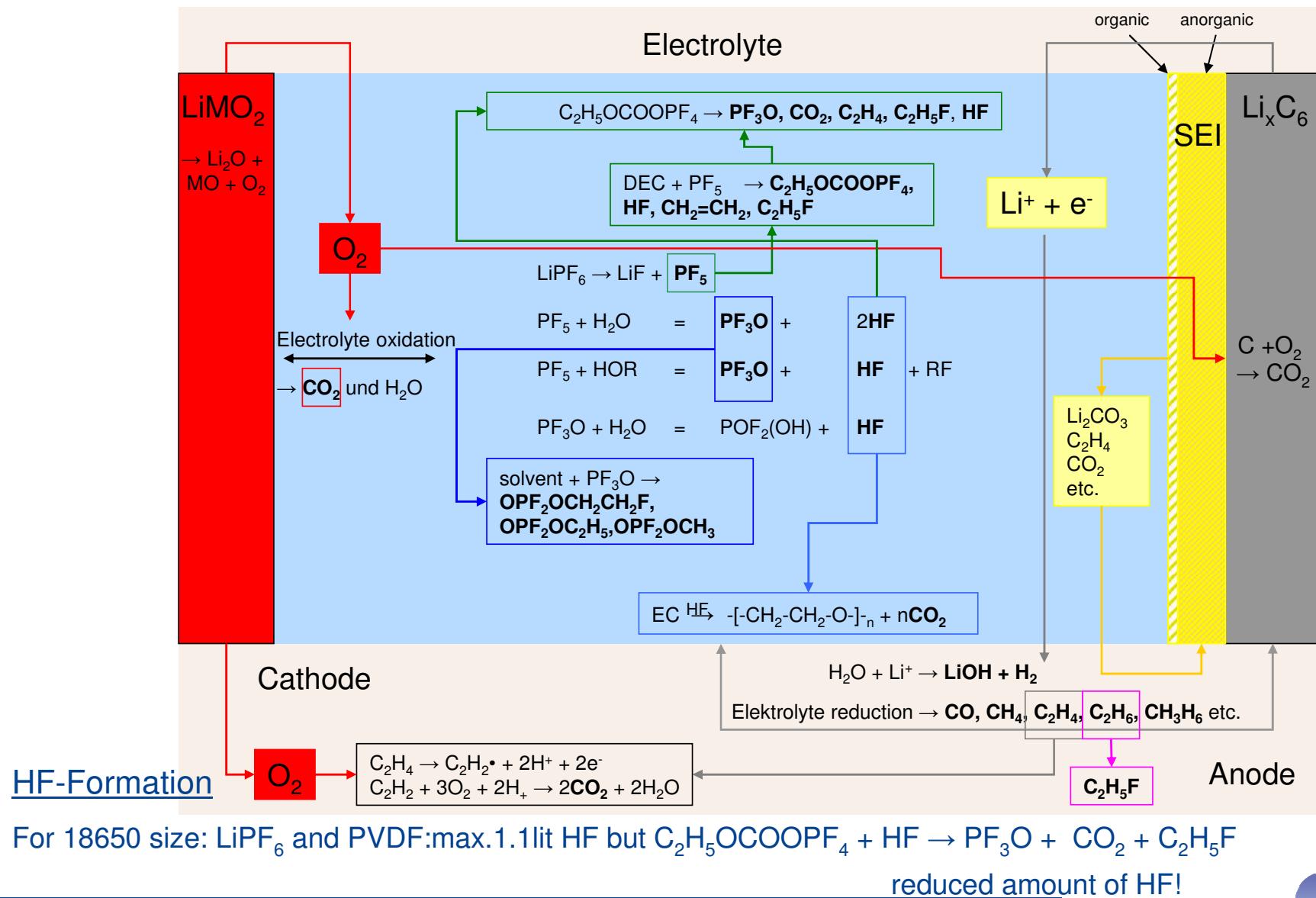
- Liquid electrolytes, organic solvents
- Polymer electrolytes
- Solid electrolytes

Flammable electrolytes, solvents

Name	Techn. Abkürzung	Siede- temperatur °C	Flammpunkt °C
Ethylencarbonat	EC	248	160
Propylencarbonat	PC	242	135
Dimethylcarbonat	DMC	90	15
Diethylcarbonat	DEC	127	33
Ethylmethylcarbonat	EMC	108	23

self ignition @ around 450 °C

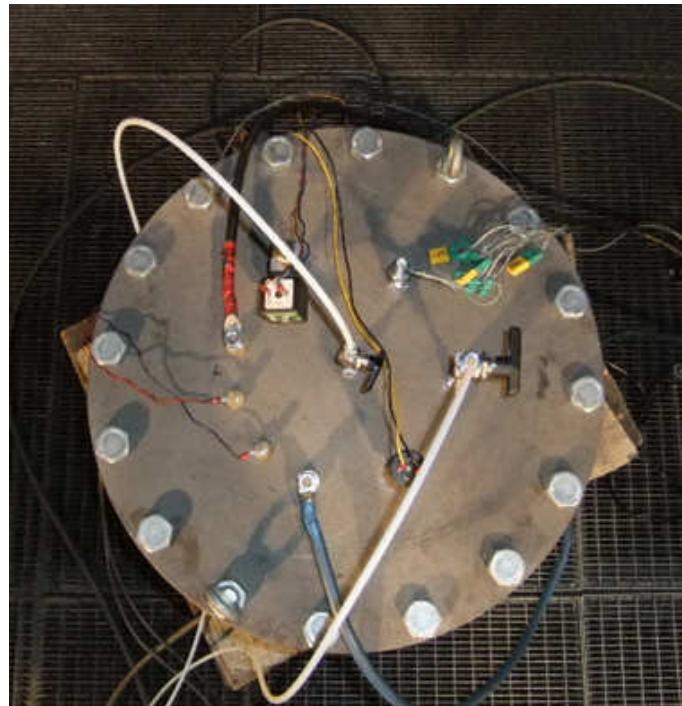
# Decomposition Reactions and Gas Formation of Battery Components



For 18650 size:  $\text{LiPF}_6$  and PVDF: max. 1.1 lit HF but  $\text{C}_2\text{H}_5\text{OCOOPF}_4 + \text{HF} \rightarrow \text{PF}_3\text{O} + \text{CO}_2 + \text{C}_2\text{H}_5\text{F}$

reduced amount of HF!

# Gas composition after overcharge different scenarios

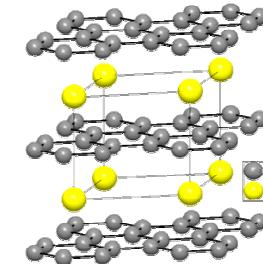


Beispiel für die Gaszusammensetzung bzw. Stoffemission nach einem "Event" bei 1C Überladung für eine 50Ah Zelle			
	Einheit	Zelle öffnet bei ca. 100°C ohne thermal runaway	thermal runaway T ca. 600°C
emittierte Gasmenge	lit	11	140
Fluorid angenommen aus HF	mg	10.1	500.6
O2	lit	9.2	0.7
N2	lit	36.0	28.9
H2	lit	<0.05	52.4
CO2	lit	3.28	51.5
CO	lit	<0.05	14.4
Phosphin	µg	1.3	530
Formaldehyd	µg	90	<20
Acetaldehyd	µg	2000	15000
Propionaldehyd	µg	170	6000
Butyraldehyd	µg	20	3800
Valeraldehyd	µg	<10	400
Methan	ml	230	2
Ethan	ml	-	1
Ethen	ml	150	8400
Propan	ml	10	1200
Propen	ml	<1	2700

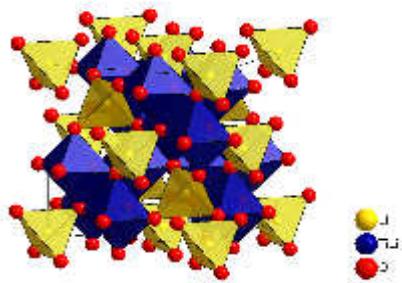
# Anodes

## C-based Anode

- source of heat
- Heat generation already at low temperatures
- High stability of the SEI is the key to a stable anode
- exfoliation > 250° C (in the presence of electrolyte)



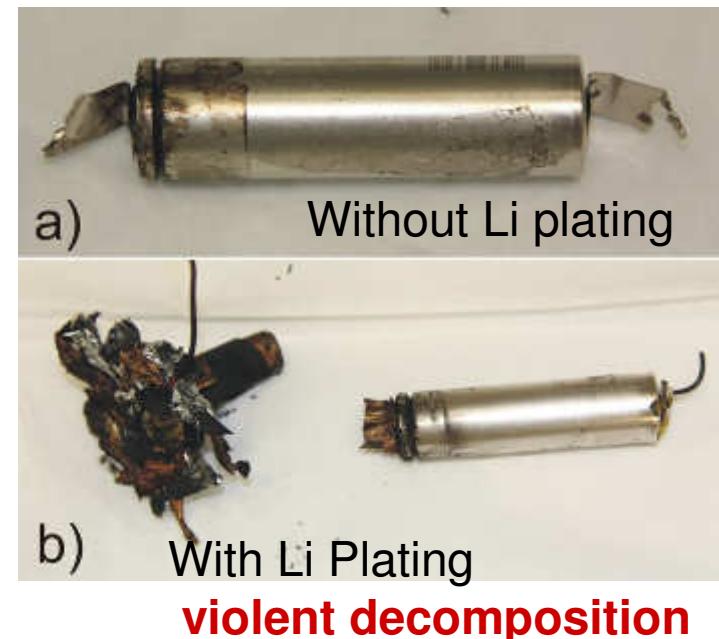
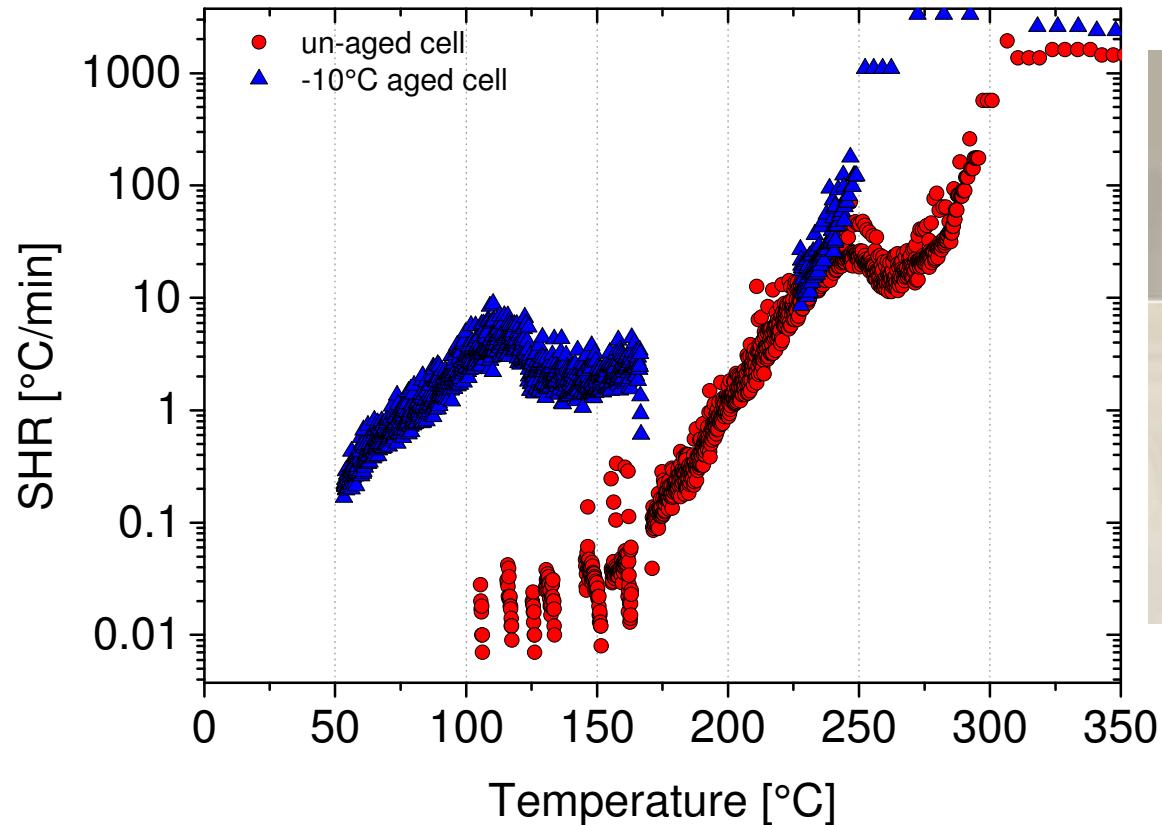
- Stability influenced by:
  - particle shape and size
  - chemistry of the electrolyte



Alternative: Titanate (more positive potential, lower cell voltage)  
high thermal stability (~1000°C)



# Lithium plating: a real safety issue! ARC run of LMO/NMC-Blend Graphite 18650 Cells



Li-plated cells: heat formation is shifted to lower temperatures and higher SHR

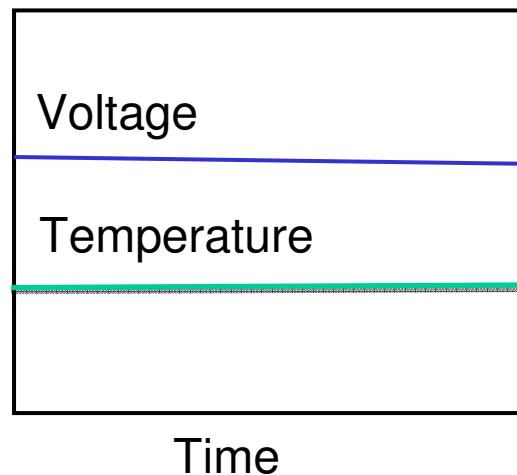
# Internal short circuit is one of failure modes of Li-ion cells that is not externally controllable

They might have different sources:

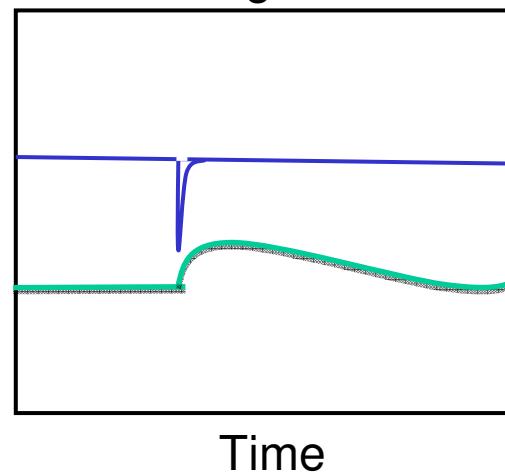
- formation of Li or Cu dendrites
- mechanical failures like crack or pin holes in the separator
- penetration by particles
- mechanical impacts and others



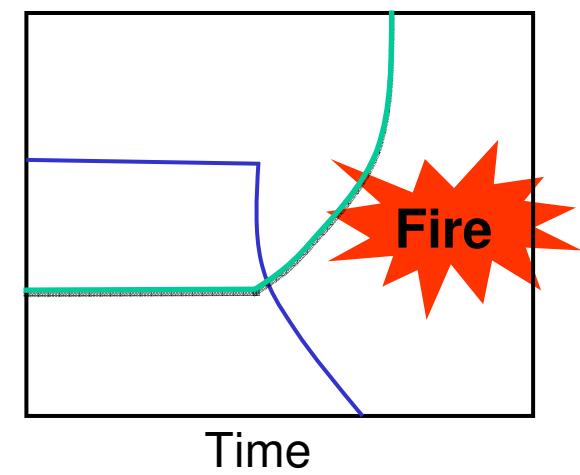
No visible impact



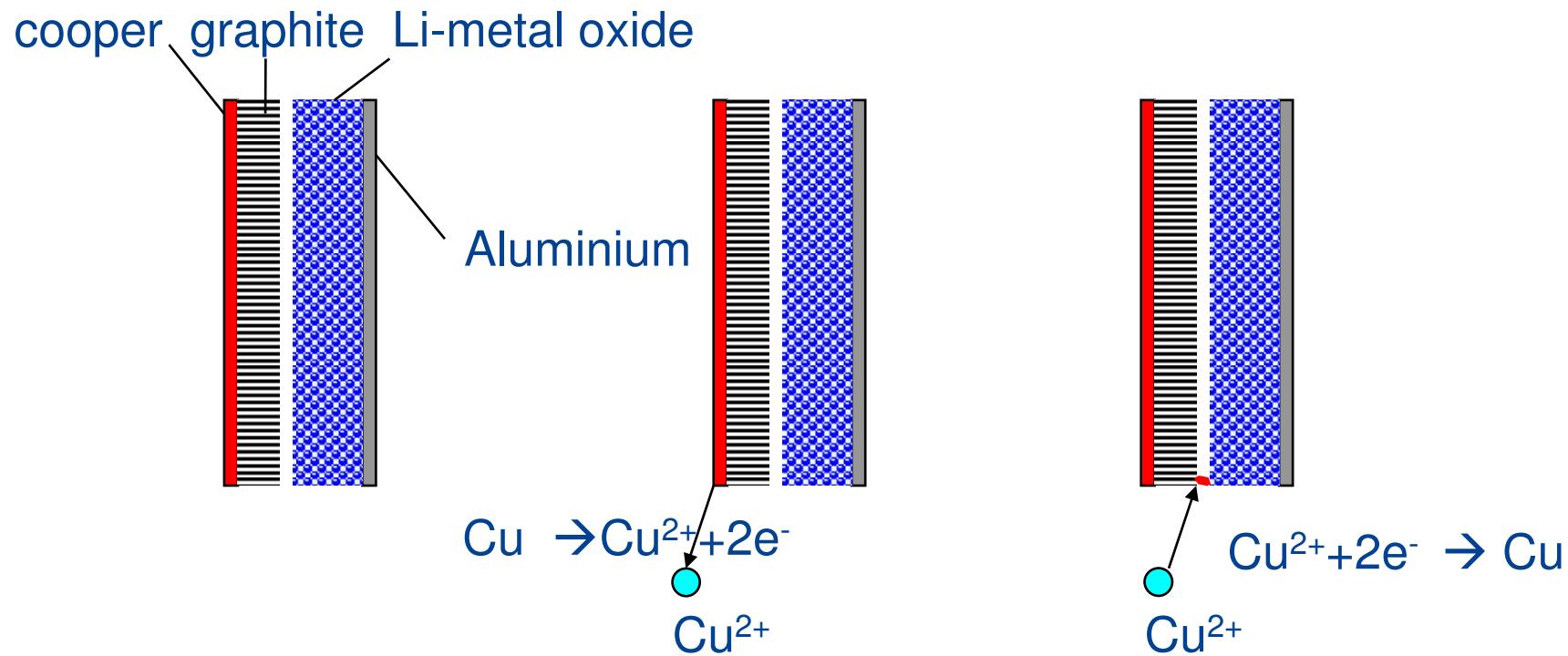
Heat generation,  
discharge



Thermal runaway



# Over-discharge dissolution of Cu followed by formation of Cu dendrites

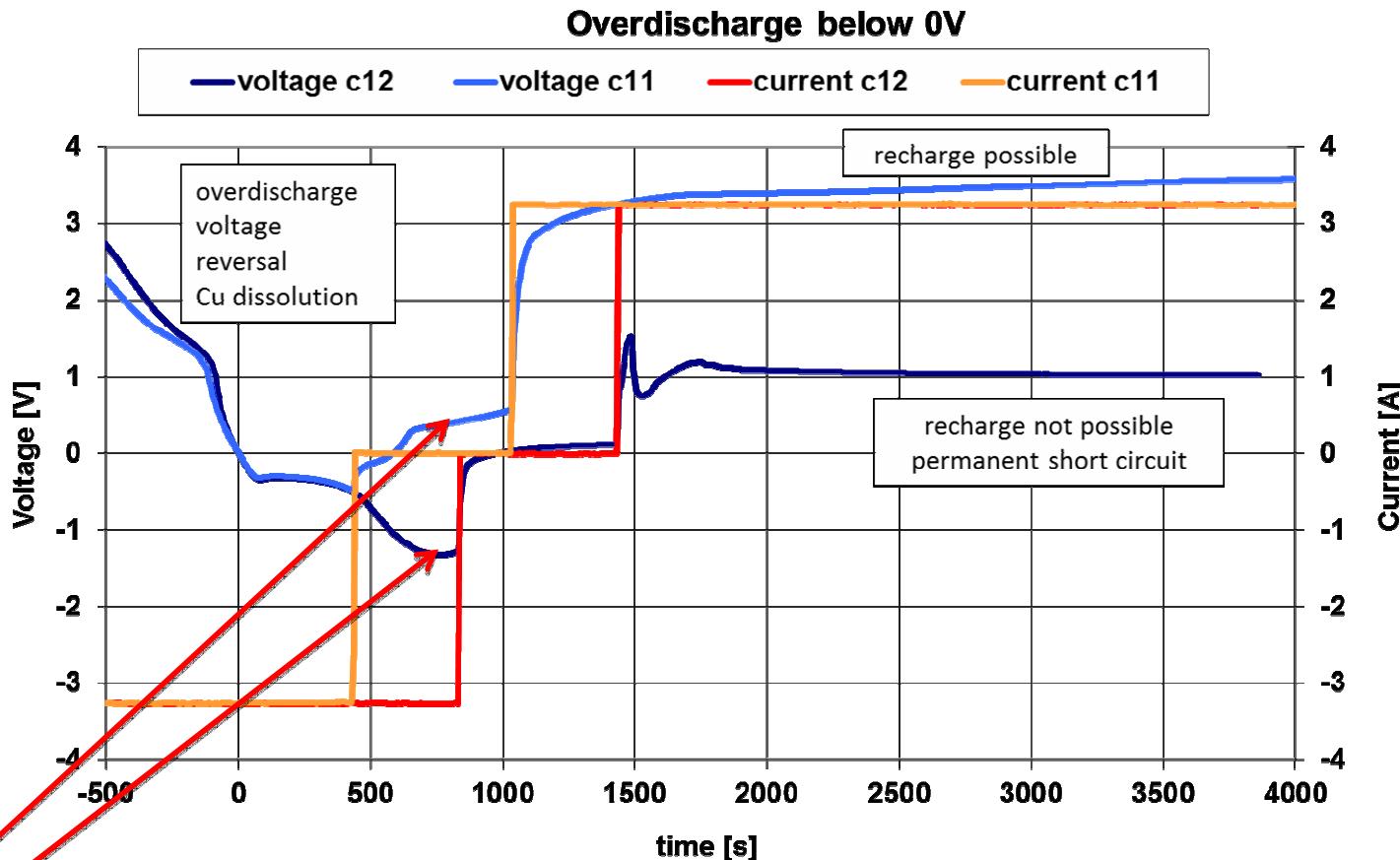


Normal operation

Deep discharge  
over discharge

Charge after  
deep discharge

# Cu dendrites in result of deep discharge Cu dissolution and deposition



Deep discharge → Cu-corrosion → recharge → Cu deposition anode → short circuit possible  
Deep discharge → Cu-corrosion → Cu deposition cathode → short circuit, no recharge possible

# Nail penetration / Puncture

## SANDIA/SAEJ 2464:

cell: Ø 3 mm, mild steel (conductive), perpendicular to electrode plates, through cell,  
80mm/sec (<250 mm/s)

module: Ø 20 mm, trough 3 cells or 100 mm

**QCT 743:** high-temperature resistant, steel spike of Ø 3 mm ~ Ø 8 mm through cell,  
at the rate of 10 mm/s ~ 40 mm/s (with the spike retained in the cell)

module: trough 3 cells

Additional influenced by:

tip shape, surface of the nail, alloy composition, .....

Very strong impact

- leaves space for variation of test conditions
- limits reproducibility

Therefore nail penetration test is disliked,  
but still the test carried out most frequently.



# Nail penetration



Nail ST37 cell without plates → hazard level 6 (fire, rupture)

# Failure Propagation

- Single cell suffered from severe failure (thermal runaway)
- Cell generate heat
- Heat transferred to the neighboring cells by heat conductivity, flames, hot gases, ...
- High probability of failure transfer to the neighboring cells causing thermal runaway of the neighbors /complete pack

## How to prevent propagation

- Safety distance between cells
- Thermal barriers between cells
- Heat sinks extracting/consuming the heat energy



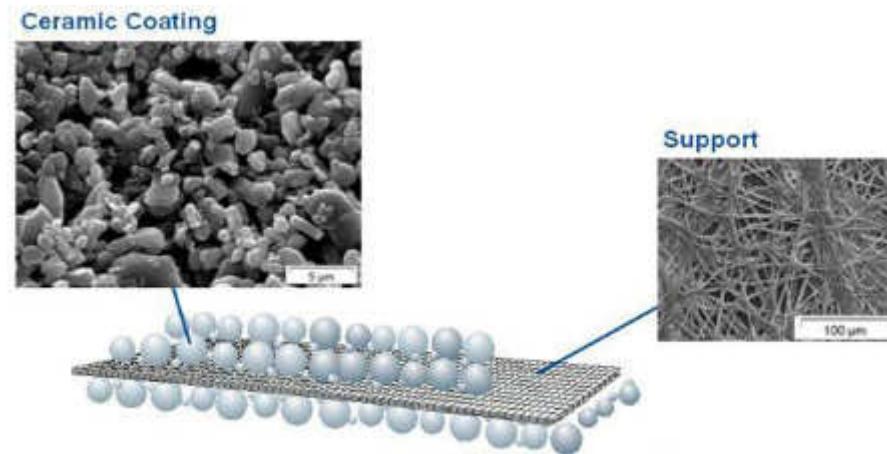
Hersman, NTSB, Investigative Update of Battery Fire Japan Airlines B-787 - Jan 7, 2013

# Safety enhancing components

## Separator

### Ceramic coated Separator

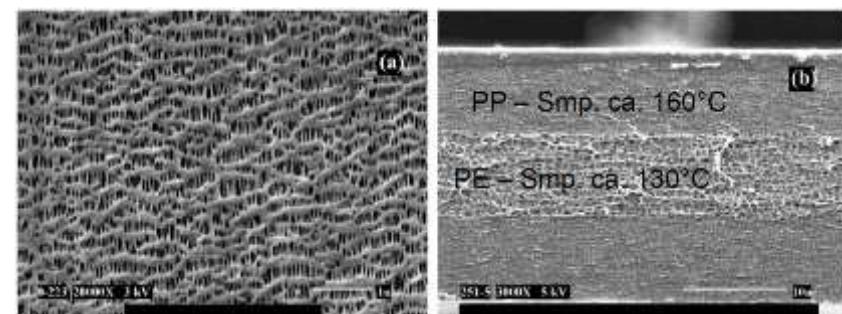
- Ceramic coating still separate the electrode even if the carrier melted away, extreme thermal stable
- Conventional polyolefin separator, melting shrinkage direct contact of the electrodes possible



Quelle: Evonik Separation

### Shut-down-Separator (3 layer)

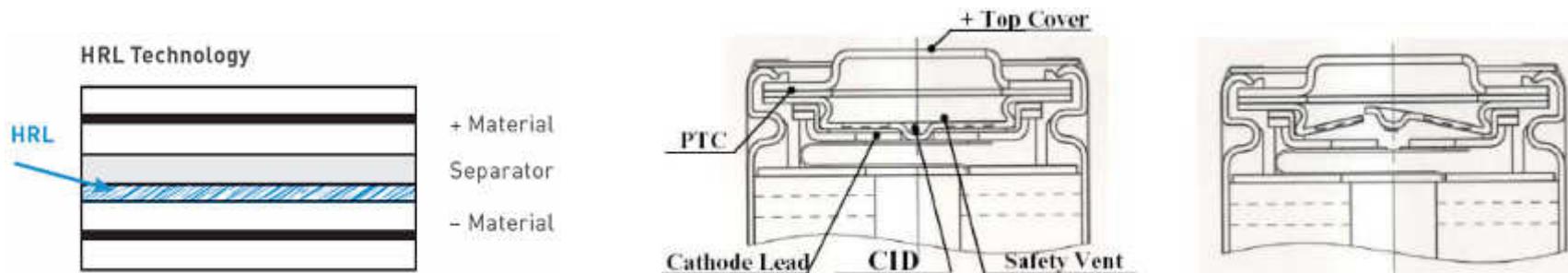
- Inner layer melting, closed pores,
- Lost Ionic conductivity
- Current path way interrupted



Quelle: Battery Separators. P Arora, ZJ Zhang, Celgard.Chem. Rev. 2004, 104, 4419-4462

# Safety enhancing components

- Heat Resistive Layer, HRL, reduce interaction of material-electrolyte
- Burst membrane release accumulated gases before cell case rupture
- Current Interruption Device (CID)
- Overcharge safety device (pressure activated, internal fuse interruption)
- Over current protection Positive Temperature Coefficient Device ( PTC)

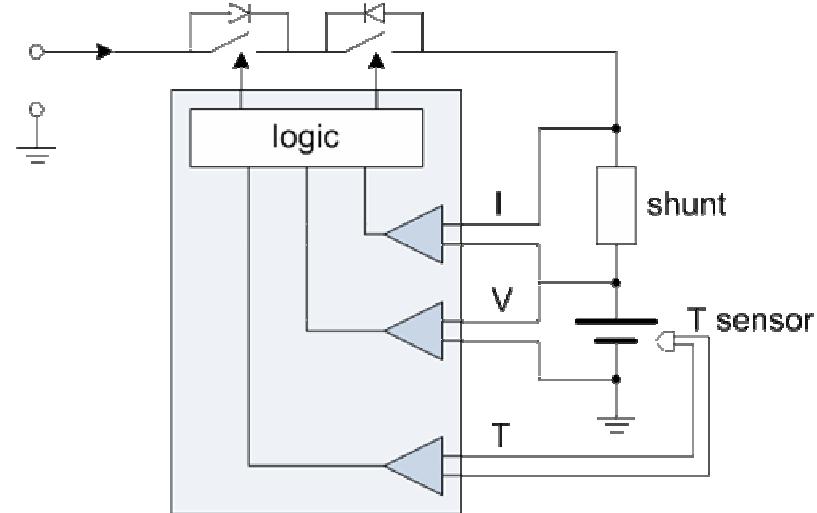


# BMS

- Monitoring
- Current/power control or interruption
- Balancing
- Thermal management
- Isolation resistance
- SOC, SOH, SOF

## Preventing/Limitation

- Overcharge
- Deep discharge/ over discharge
- Short circuit
- Overcurrent/ overpower
- Over temperatures



## Critical parameters: redundancy

- E.g. cell voltage-module voltage-cell voltage monitoring and control independent voltage comparator

# Fazit

## Lithium-Ion-Technology

- available technology
- no intrinsic safe cell technology
- suitable controllable by BMS
- Safe by
  - Selecting suitable materials and additives
  - Use of safety and protection components
  - Design of cell and module
  - suitable control by BMS
  - Monitoring and control of current, voltage, temperature
  - Redundancy for critical parameters
  - Protection against re-initializing after failure
  - Suitable housing/hosting of the battery

# Thank you for your Attention !

## Acknowledgement:

Thanks to my colleagues and co-workers of the battery group  
and the material group of ZSW-Ulm

- **SPEISI: Sicherheit und Zuverlässigkeit von PV-Anlagen mit Speichersystemen unter besonderer Berücksichtigung von Brandrisiken und Löschstrategien**  
FKZ: 41V6971



Bundesministerium  
für Umwelt, Naturschutz  
und Reaktorsicherheit

- **Sicherheit und Netzdienlichkeit von elektrischen Heimspeichersystemen mit Lithium-Ionen-Batterien**  
FKZ: 03ET6055C



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